

MATERIAL SAFETY DATA SHEET

ASBURY FLUXMASTER OF UTAH
2855 Franklin Canyon Road
Rodeo, CA 94572

#58cur

Rev. Date 3/1/96
Emergency Telephone
800-424-9300

PRODUCT IDENTIFICATION

Chemical name: # 751, 752, 753, 756, 759 Degasser
Chemical Family: Mixture of Organic and Inorganic salts and
oxides
Formula: Mixture
Trade name: Flux salts

HAZARDOUS CONSTITUENTS

| % of Mix | Name | CAS # | Niosh | OSHA Pel | ACGIH tlv |
|----------|---------------------|------------|-------|----------------------------|-----------|
| 30-80 | Hexachloroethane | 62-72-1 | | 10 | 10 79 |
| 20-70 | Potassium Chloride | 7447-40-7 | | n/l ¹ 15*;5** | 10 |
| 0-20 | Sodium Fluosilicate | 16893-85-9 | | none 2.5 (F ⁻) | 2.5 (F) |

values expressed in Mg/m³

PHYSICAL PROPERTIES

Freezing point: N/A
Melting point: approx. 990 C
Boiling point:
Sublimes:
Evaporation: N/A
Appearance and odor: White to tan granular solid or tablet. Odor of naptha.

Vapor Pressure: N/A
Vapor Density: .08mm Hg
Density (g/cc): 2.9
H₂O Solubility: Insoluable
% Volatiles : none

EXPLOSION AND REACTIVITY DATA

Flash Point: none
Flammable (explosive) Limits: V/V%
LEL: none
UEL: none

Extinguishing Media: This material is non-combustible. Use extinguishing media suitable for the surrounding fire.

Special Firefighting Procedures: If this material is involved in a fire-fighting situation, use a self-contained breathing in pressure demand mode.

Unusual Fire or Explosion Hazards: No unusual explosion hazards known.

General Reactivity Information: The product is a stable material.

Incompatibility: Contact with acid may liberate hydrogen fluoride gas which is toxic and corrosive and or chlorine fumes.

Hazardous Decomposition Products: During melting operations and at elevated temperatures, fluoride compounds will be liberated, including, but not limited to hydrogen fluoride and potassium fluoride and chlorine fumes.

HEALTH HAZARD INFORMATION/EMERGENCY PROCEDURES:

General: Exposure to dust or fumes of fluoride containing salts may present significant health hazards. Fluoride salts may cause acute poisoning and/or death (principally by ingestion). Skin contact and exposure to soft tissues or mucous membranes may result in severe irritation and/or tissue damage. Crippling bone changes, densification and mottling of tooth enamel are reported as chronic effects of overexposure. These effects are not common currently in industrial employees.

ACUTE- Ingestion of fluoride salts can cause gastric pain, internal bleeding, tissue damage and death. Nose bleeds, skin rashes, eye irritation and slow healing scars may result if exposure is excessive. Fluoride salts are irritating to the skin and mucous membranes.

CHRONIC- Chronic exposure may lead to calcification of the bones and ligaments, osteosclerosis. Prolonged exposure to fluoride salts may cause tissue damage.

Inhalation: NIOSH approved respirators are recommended if engineering controls are not feasible or unable to maintain a concentration below that specified. If overexposed, remove victim to fresh air. Rinse mouth and nasal passages with water if the person is conscious.

Ingestion: Do not induce vomiting in an unconscious subject. Seek medical attention immediately.

Dermal: Wear gloves if prolonged or repeated contact is expected. If irritation is evident, wash the contaminated area repeatedly with water and a mild soap. Always wash thoroughly after contact with fluoride salts.

Ocular: Proper protective equipment is recommended at all times. Safety glasses should be worn at any time there is a reasonable probability of exposure. In the event of exposure, flush the eyes with large amounts of water, occasionally lifting the upper and lower lids. Seek medical attention as burns may not be evident or may appear later.

Carcinogenic References: None of the materials used in this product have been listed as carcinogens by the suppliers. Additionally, none of the materials have been found to be listed as carcinogens the NIOSH guide.

ENVIRONMENTAL PROTECTION INFORMATION:

In the event of a spill of powder or dust, clean-up should be conducted using a vacuum system with a high efficiency particulate air filtration system. Caution should be taken to minimize airborne generation of powder or dust and to avoid contamination of air and water.

Fluoride compounds may have significant impact on air and water quality. Airborne emissions, spills and releases to the environment (discharge to streams, sewer systems, ground water, surface soil, etc.) should be controlled immediately. If such potential for a spill or release exists, it is advisable to develop an emergency spill response plan. It is also advisable to consider monitoring ambient air as well as any effluent which may contain fluorides if potential exists for damage to aquatic or terrestrial ecosystems.

ENVIRONMENTAL PROTECTION INFORMATION (cont.) :

State or federal regulations may require specific labeling, packaging, storage, transportation and disposal procedures. It is recommended to contact an Environmental Engineer or consultant familiar with the applicable waste disposal regulations.

SPECIAL PRECAUTIONS:

When working with fluoride containing salts it is advisable to wear eye protection, Niosh approved respirator, and gloves to limit exposure. Contact with the actual material should be minimized. Always wash thoroughly prior to eating or smoking.

This product must be handled accordingly to the size, shape and quantity of material involved. Drums may require use of hoists, cranes etc. Store this material in a dry area. Fluoride salts should not be stored adjacent to acids. Keep away from contact with food or feed products.

DOT SHIPPING INFORMATION:

This material is not regulated by the DOT for land transportation. The following is the most accurate description of the material for shipping purposes.

"CHEMICALS, NOS"